Table 1. Selected geometric parameters $\left(\AA^{\circ},^{\circ}\right)$

| $\mathrm{S}-\mathrm{O} 2$ | $1.439(3)$ | $\mathrm{C} 1-\mathrm{C} 5$ | $1.575(5)$ |
| :--- | :---: | :--- | ---: | :--- |
| $\mathrm{S}-\mathrm{O} 1$ | $1.442(3)$ | $\mathrm{C} 2-\mathrm{C} 3$ | $1.321(5)$ |
| $\mathrm{S}-\mathrm{C} 10$ | $1.777(3)$ | $\mathrm{C} 5-\mathrm{C} 9$ | $1.522(5)$ |
| $\mathrm{S}-\mathrm{C} 1$ | $1.803(3)$ | $\mathrm{C} 5-\mathrm{C} 6$ | $1.555(5)$ |
| $\mathrm{C} 1-\mathrm{C} 7$ | $1.542(4)$ | $\mathrm{C} 6-\mathrm{C} 7$ | $1.533(5)$ |
| $\mathrm{O} 2-\mathrm{S}-\mathrm{O} 1$ | $119.0(2)$ | $\mathrm{C} 7-\mathrm{C} 1-\mathrm{S}$ | $116.7(2)$ |
| $\mathrm{O} 2-\mathrm{S}-\mathrm{C} 10$ | $105.6(2)$ | $\mathrm{C} 5-\mathrm{C} 1-\mathrm{S}$ | $123.6(2)$ |
| $\mathrm{O} 1-\mathrm{S}-\mathrm{C} 10$ | $107.4(2)$ | $\mathrm{C} 9-\mathrm{C} 5-\mathrm{C} 4$ | $113.5(3)$ |
| $\mathrm{O} 2-\mathrm{S}-\mathrm{Cl}$ | $105.7(2)$ | $\mathrm{C} 9-\mathrm{C} 5-\mathrm{C} 6$ | $112.5(3)$ |
| $\mathrm{O} 1-\mathrm{S}-\mathrm{C} 1$ | $107.8(2)$ | $\mathrm{C} 4-\mathrm{C} 5-\mathrm{C} 6$ | $114.8(3)$ |
| $\mathrm{C} 10-\mathrm{S}-\mathrm{C} 1$ | $111.28(14)$ | $\mathrm{C} 9-\mathrm{C} 5-\mathrm{C} 1$ | $121.2(3)$ |
| $\mathrm{C} 2-\mathrm{Cl}-\mathrm{C} 7$ | $114.9(3)$ | $\mathrm{C} 4-\mathrm{C} 5-\mathrm{Cl}$ | $106.4(3)$ |
| $\mathrm{C} 2-\mathrm{Cl}-\mathrm{C} 5$ | $103.0(3)$ | $\mathrm{C} 6-\mathrm{C} 5-\mathrm{Cl}$ | $85.9(3)$ |
| $\mathrm{C} 7-\mathrm{Cl}-\mathrm{C} 5$ | $90.5(3)$ | $\mathrm{C} 7-\mathrm{C} 6-\mathrm{C} 5$ | $91.6(3)$ |
| $\mathrm{C} 2-\mathrm{Cl}-\mathrm{S}$ | $107.3(2)$ | $\mathrm{C} 6-\mathrm{C} 7-\mathrm{Cl}$ | $87.9(2)$ |

H atoms were refined with fixed geometry, each riding on a carrier atom, with an isotropic displacement parameter 1.5 (for methyl H atoms) or 1.2 (for the other H atoms) times the value of the equivalent isotropic displacement parameter of the carrier atom.

Data collection: CAD-4 Software (Enraf-Nonius, 1989). Cell refinement: CAD-4 Software. Data reduction: MolEN (Fair, 1990). Program(s) used to solve structure: SHELXS86 (Sheldrick, 1990). Program(s) used to refine structure: SHELXL93 (Sheldrick, 1993). Molecular graphics: ZORTEP (Zsolnai, 1995). Software used to prepare material for publication: SHELXL93.

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# The Antifungal Drug Clotrimazole 

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#### Abstract

The structure of the title compound, 1-[(2-chloro-phenyl)diphenylmethyl]-1 H -imidazole, $\mathrm{C}_{22} \mathrm{H}_{17} \mathrm{ClN}_{2}$, has been determined. The molecular conformation showed a weathercock-type structure and the three phenyl rings are almost perpendicular to the imidazole ring. The distances between the centres of the three phenyl rings and the centre of the imidazole ring are in the range 4.524.54 Å.


## Comment

The title compound, (I) (common names Clotrimazole, Lotromin, Mycelex-G and Canesten), is a member of antifungal imidazole derivatives with broad-spectrum activity against tinea infections and candidiasis (Hoogerheide \& Wyka, 1982). It is generally believed that antifungal activity of imidazoles results from the disorganization of the fungal cytoplasmic membrane, which is a consequence of inactivation of P-450 by binding between the N atom of the imidazole ring and the haem site (Hansch et al., 1990).

(I)

Recently, the mechanism of antifungal agents has been clarified by X-ray analysis studies on inhibitors bound to P-450 (Poulous, Finzel et al., 1985, 1987; Poulous \& Howard, 1987). Crystal structures of antifungal imidazoles have also been reported (Peeters et al., 1979; Freer et al., 1986; Shin et al., 1987).

As part of our studies on the structure-activity relationship of its imidazole agents, the crystal structure of (I) has been determined. The title compound was purchased from SIGMA chemicals.

The overall molecular conformation of the title compound may be described as a weathercock type which
has the three phenyl rings as the fan. The three phenyl rings are almost perpendicular to the $\mathrm{C} 7-\mathrm{N} 20$ bond axis of the imidazole ring. The dihedral angles between the three phenyl rings ( $A, B$ and $C$ ) and the imidazole ring are $126.8(3), 78.5(3)$ and $107.5(2)^{\circ}$, respectively. Each of the four ring systems in the molecule is almost planar, the maximum deviation of 0.010 (2) $\AA$ is at the C14 and C15 atoms of phenyl ring $C$. In the imidazole ring the $\mathrm{C} 21=\mathrm{N} 22$ and $\mathrm{C} 23=\mathrm{C} 24$ bond lengths are 1.315 (3) and 1.346 (3) $\AA$, respectively. These values are much shorter ( $0.032 \AA$ about C23-C24) than those of pure imidazole ( $1.378 \AA$; McMullan et al., 1979). In the three phenyl rings, the $\mathrm{C}=\mathrm{C}$ bond lengths range from 1.365 (4) to 1.400 (3) $\AA$ and the average distance is $1.381 \AA$. These values are in good agreement with those in miconazole (Peeters et al., 1979), econazole (Freer et al., 1986), econazole nitrate (Suh et al., 1990) and sulconazole nitrate (Shin et al., 1997).


Fig. 1. Molecular structure showing $40 \%$ probability displacement ellipsoids.

## Experimental

Crystals of (I) were grown from chloroform solution by slow evaporation at room temperature.

## Crystal data

$\mathrm{C}_{22} \mathrm{H}_{17} \mathrm{ClN}_{2}$
$M_{r}=344.83$
Triclinic
$P \overline{1}$
$a=8.7590(5) \AA$
$b=10.5540$ (12) $\AA$
$c=10.6064(14) \AA$
$\alpha=114.116(14)^{\circ}$
$\beta=96.957(7)^{\circ}$
$\gamma=97.535(7)^{\circ}$
$V=870.5(2) \AA^{3}$
$Z=2$
$D_{x}=1.316 \mathrm{Mg} \mathrm{m}^{-3}$
$D_{m}$ not measured
Data collection
CAD-4 diffractometer
$\omega-2 \theta$ scans
Absorption correction: none
3177 measured reflections
2967 independent reflections
2345 reflections with
$I>2 \sigma(I)$
$R_{\mathrm{int}}=0.010$

## Refinement

Refinement on $F^{2}$
$R(F)=0.037$
$w \cdot R\left(F^{2}\right)=0.086$
$S=1.06$
2967 reflections
294 parameters
All H -atom parameters refined

Mo $K \alpha$ radiation
$\lambda=0.71069 \AA$
Cell parameters from 25 reflections
$\theta=4.5-14.5^{\circ}$
$\mu=0.225 \mathrm{~mm}^{-1}$
$T=293(2) \mathrm{K}$
Prism
$0.42 \times 0.25 \times 0.21 \mathrm{~mm}$
Colourless
$\theta_{\text {max }}=25^{\circ}$
$h=0 \rightarrow 9$
$k=-11 \rightarrow 11$
$l=-11 \rightarrow 11$
3 standard reflections every 200 reflections frequency: 60 min intensity decay: $<2 \%$

$$
\begin{aligned}
& \mathrm{u}^{\prime}=1 /\left[\sigma^{2}\left(F_{o}\right)^{2}+(0.04 P)^{2}\right. \\
& \quad+0.3261 P] \\
& \text { where } P=\left(F_{o}^{2}+2 F_{c}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }=0.001 \\
& \Delta \rho_{\max }=0.23 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-0.22 \mathrm{e} \AA^{-3} \\
& \text { Extinction correction: none } \\
& \text { Scattering factors from } \\
& \text { International Tables for } \\
& \text { Crystallography (Vol. C) }
\end{aligned}
$$

To obtain information for use in structure-activity studies of antifungal imidazoles, the distances ( $l_{1}, l_{2}$ and $l_{3}$ ) between the centre of the imidazole ring and the centres of the three aromatic rings $(A, B$ and $C$ ) were calculated. The distances $l_{1}, l_{2}$ and $l_{3}$ are $4.52,4.54$ and $4.52 \AA$, respectively. The corresponding distance $l_{1}$ in the antifungal imidazoles such as miconazole, econazole and sulconazole nitrate are within the range $4.52-4.97 \AA$ (except for econazole nitrate, $5.70 \AA$ ).

The dihedral angles between the aromatic ring $A$, which is halogen substituted, and the imidazole ring are in the range $95.2-126.8^{\circ}$ (except for econazole nitrate, $10.6^{\circ}$ ).

The result of conformational analysis may be used as a basis for the definition of structural parameters necessary for the range of activity of this class of compounds.

Table 1. Selected geometric parameters $\left(\AA,^{\circ}\right)$

| $\mathrm{Cl}-\mathrm{C} 1$ | $1.743(2)$ | $\mathrm{C} 10-\mathrm{C} 11$ | $1.370(3)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{Cl}-\mathrm{C} 2$ | $1.381(3)$ | $\mathrm{C} 1-\mathrm{C} 12$ | $1.367(3)$ |
| $\mathrm{C} 1-\mathrm{C} 6$ | $1.400(3)$ | $\mathrm{C} 12-\mathrm{C} 13$ | $1.376(3)$ |
| $\mathrm{C} 2-\mathrm{C} 3$ | $1.375(3)$ | $\mathrm{C} 14-\mathrm{C} 19$ | $1.379(3)$ |
| $\mathrm{C} 3-\mathrm{C} 4$ | $1.373(4)$ | $\mathrm{C} 14-\mathrm{C} 15$ | $1.395(3)$ |
| $\mathrm{C} 4-\mathrm{C} 5$ | $1.386(3)$ | $\mathrm{C} 15-\mathrm{C} 16$ | $1.374(3)$ |
| $\mathrm{C} 5-\mathrm{C} 6$ | $1.390(3)$ | $\mathrm{C} 16-\mathrm{C} 17$ | $1.382(3)$ |
| $\mathrm{C} 6-\mathrm{C} 7$ | $1.551(3)$ | $\mathrm{C} 17-\mathrm{C} 18$ | $1.365(4)$ |
| $\mathrm{C} 7-\mathrm{N} 20$ | $1.498(2)$ | $\mathrm{C} 18-\mathrm{C} 19$ | $1.379(3)$ |
| $\mathrm{C} 7-\mathrm{C} 14$ | $1.541(2)$ | $\mathrm{N} 20-\mathrm{C} 21$ | $1.356(3)$ |
| $\mathrm{C} 7-\mathrm{C} 8$ | $1.384(2)$ | $\mathrm{N} 20-\mathrm{C} 24$ | $1.371(3)$ |
| $\mathrm{C} 8-\mathrm{C} 9$ | $1.386(3)$ | $\mathrm{C} 21-\mathrm{N} 22$ | $1.315(3)$ |
| $\mathrm{C} 8-\mathrm{C} 13$ | $1.389(3)$ | $\mathrm{C} 23-\mathrm{C} 23$ | $1.372(3)$ |
| $\mathrm{C} 9-\mathrm{C} 10$ | $121.9(2)$ | $\mathrm{C} 11-\mathrm{C} 10-\mathrm{C} 9$ | $1.346(3)$ |
| $\mathrm{C} 2-\mathrm{C} 1-\mathrm{C} 6$ | $116.0(2)$ | $\mathrm{C} 12-\mathrm{C} 11-\mathrm{C} 10$ | $120.3(2)$ |
| $\mathrm{C} 2-\mathrm{C} 1-\mathrm{Cl}$ | $122.1(2)$ | $\mathrm{C} 11-\mathrm{C} 12-\mathrm{C} 13$ | $119.2(2)$ |
| $\mathrm{C} 6-\mathrm{C} 1-\mathrm{Cl}$ | $120.0(2)$ | $\mathrm{C} 12-\mathrm{C} 13-\mathrm{C} 8$ | $120.8(2)$ |
| $\mathrm{C} 3-\mathrm{C} 2-\mathrm{C} 1$ | $119.7(2)$ | $\mathrm{C} 19-\mathrm{C} 14-\mathrm{C} 15$ | $117.2(2)$ |
| $\mathrm{C} 4-\mathrm{C} 3-\mathrm{C} 2$ | $120.1(2)$ | $\mathrm{C} 19-\mathrm{C} 14-\mathrm{C} 7$ | $122.8(2)$ |
| $\mathrm{C} 3-\mathrm{C} 4-\mathrm{C} 5$ | $121.9(2)$ | $\mathrm{C} 15-\mathrm{C} 14-\mathrm{C} 7$ | $119.7(2)$ |
| $\mathrm{C} 4-\mathrm{C} 5-\mathrm{C} 6$ |  |  |  |


| $\mathrm{C} 5-\mathrm{C} 6-\mathrm{Cl}$ | $116.5(2)$ | $\mathrm{C} 16-\mathrm{C} 15-\mathrm{Cl} 14$ | $121.3(2)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{C} 5-\mathrm{C} 6-\mathrm{C} 7$ | $120.8(2)$ | $\mathrm{C} 15-\mathrm{C} 16-\mathrm{C} 17$ | $120.7(2)$ |
| $\mathrm{Cl}-\mathrm{C} 6-\mathrm{C} 7$ | $122.7(2)$ | $\mathrm{C} 8-\mathrm{C} 17-\mathrm{C} 16$ | $118.3(2)$ |
| $\mathrm{N} 20-\mathrm{C} 7-\mathrm{Cl} 14$ | $111.17(14)$ | $\mathrm{C} 17-\mathrm{C} 18-\mathrm{C} 19$ | $121.2(2)$ |
| $\mathrm{N} 20-\mathrm{C} 7-\mathrm{C} 6$ | $106.75(14)$ | $\mathrm{C} 14-\mathrm{C} 19-\mathrm{C} 18$ | $121.4(2)$ |
| $\mathrm{C} 14-\mathrm{C} 7-\mathrm{C} 6$ | $111.50(15)$ | $\mathrm{C} 21-\mathrm{N} 20-\mathrm{C} 24$ | $106.0(2)$ |
| $\mathrm{N} 20-\mathrm{C} 7-\mathrm{C} 8$ | $108.76(14)$ | $\mathrm{C} 21-\mathrm{N} 20-\mathrm{C} 7$ | $124.7(2)$ |
| $\mathrm{C} 14-\mathrm{C} 7-\mathrm{C} 8$ | $107.43(15)$ | $\mathrm{C} 24-\mathrm{N} 20-\mathrm{C} 7$ | $129.2(2)$ |
| $\mathrm{C} 6-\mathrm{C} 7-\mathrm{C} 8$ | $111.24(15)$ | $\mathrm{N} 22 \mathrm{C} 21-\mathrm{N} 20$ | $112.5(2)$ |
| $\mathrm{C} 9-\mathrm{C} 8-\mathrm{C} 13$ | $117.4(2)$ | $\mathrm{C} 21-\mathrm{N} 22-\mathrm{C} 23$ | $104.2(2)$ |
| $\mathrm{C} 9-\mathrm{C} 8-\mathrm{C} 7$ | $122.5(2)$ | $\mathrm{C} 24-\mathrm{C} 23-\mathrm{N} 22$ | $111.1(2)$ |
| $\mathrm{C} 13-\mathrm{C} 8-\mathrm{C} 7$ | $120.2(2)$ | $\mathrm{C} 23-\mathrm{C} 24-\mathrm{N} 20$ | $106.1(2)$ |
| $\mathrm{C} 8-\mathrm{C} 9-\mathrm{C} 10$ | $121.2(2)$ |  |  |

The space group $P \overline{1}$ was determined from non-systematic absences. Intensity data were corrected for Lorentz and polarization effects. The crystal structure was solved by direct methods using SHELXS86 (Sheldrick, 1985) and refined by full-matrix least-squares methods using SHELXL93 (Sheldrick, 1993) with anisotropic displacement parameters for all nonH atoms. All H atoms were located from difference Fourier maps and were refined isotropically in the final cycles. There were no significant features in the final difference Fourier map. Refinement of the H atoms led to $\mathrm{C}-\mathrm{H}$ distances in the range 0.88 (3)-1. 02 (2) $\AA . U$ values vary from 0.038 (6) to 0.078 (8) with a mean of $0.057(8) \AA^{2}$. All geometrical calculations were performed using GEOM. All computations were performed using IBM PCs.

Data collection: CAD-4 Software (Enraf-Nonius, 1989). Cell refinement: CAD-4 Software. Data reduction: Xtal3.2 (Hall et al., 1992). Program(s) used to solve structure: SHELXS86. Program(s) used to refine structure: SHELXL93. Molecular graphics: ORTEX (McArdle, 1995). Software used to prepare material for publication: SHELXL93.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: BK1362). Services for accessing these data are described at the back of the journal.

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## (4S,5R,2'R)-3,4-Dimethyl-1-[2'-(1-naphthyl)-$2^{\prime}$-phenylacetyl]-5-phenylimidazolidin-2-one

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## Abstract

The title compound, $(4 S, 5 R)-3,4$-dimethyl-1-[(2R)-2( 1 -naphthyl)-2-phenylacetyl]-5-phenylimidazolidin-2one, $\mathrm{C}_{29} \mathrm{H}_{26} \mathrm{~N}_{2} \mathrm{O}_{2}$, has two different aryl groups at the $\alpha$-carbon which confer a stereogenic character on the site. The geometry of the molecule is stabilized by an intramolecular hydrogen bond linking the acetamide and imidazolidine groups, which gives a planar character to the core.

## Comment

Acetamides have been reported as herbicides, especially those with an $\alpha$-chloro substituent (Couderchet et al., 1986). We have been studying the families of $\alpha$-aryland $\alpha, \alpha$-diarylacetamides and both have shown interesting growth-inhibitor properties against Avena sativa and Cyperus rotundus (Palacios et al., 1995). The diaryl family has a stereogenic center when the aryl groups are different. In order to synthesize enantiomerically pure $\alpha, \alpha$-(1-naphthyl)phenylacetamide, we have recently prepared the imidazolidin-2-one derivative (Lotz et al., 1994). The reaction described (see Experimental) provides two epimeric species, viz. ( $4 S, 5 R, 2^{\prime} S$ )-, (1), and (4S,5R, $2^{\prime} R$ )-3,4-dimethyl-1-[2'-(1-naphthyl)-2'-phenyl-

